

# Chapter 7

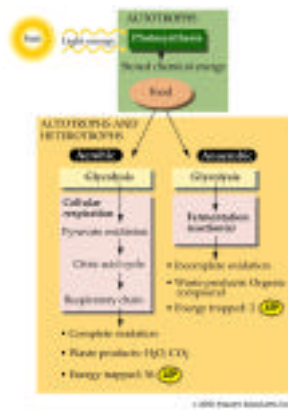
## Cellular respiration

How do organisms acquire energy?

- Inorganic materials
  - Autotrophs
- Organic materials
  - Heterotrophs

### Autotrophs and heterotrophs

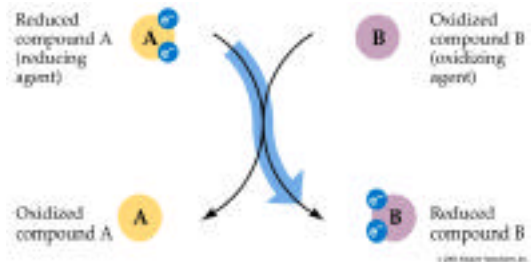
Both autotrophs and heterotrophs use **cellular respiration** to obtain energy for growth and reproduction.



### Oxidation - reduction

Transfer of electrons

oxidation = loss of electrons  
reduction = gain of electrons

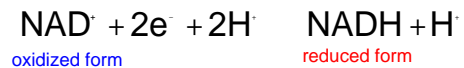


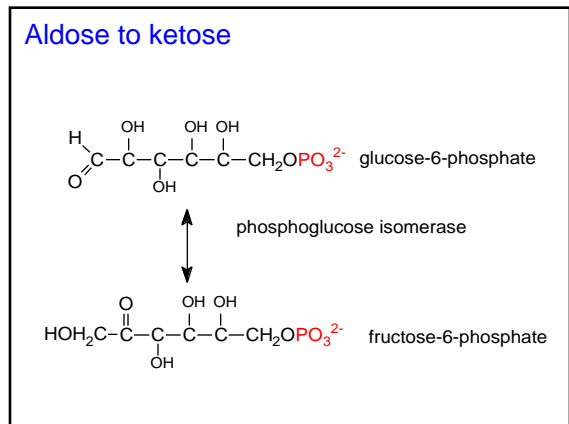
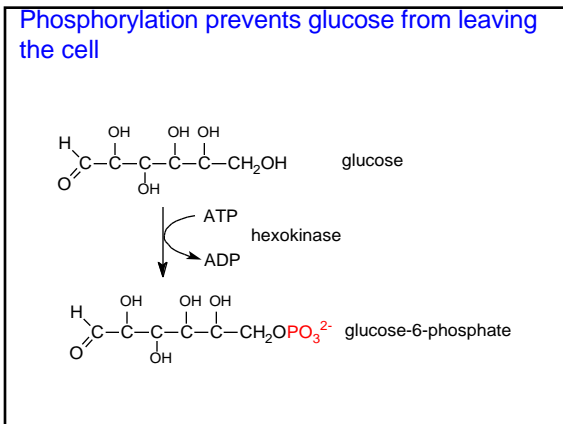
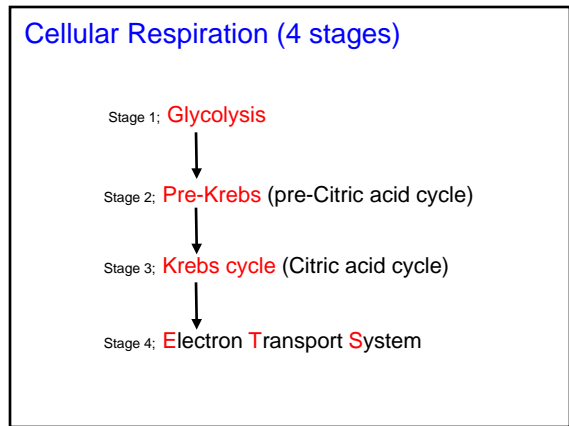
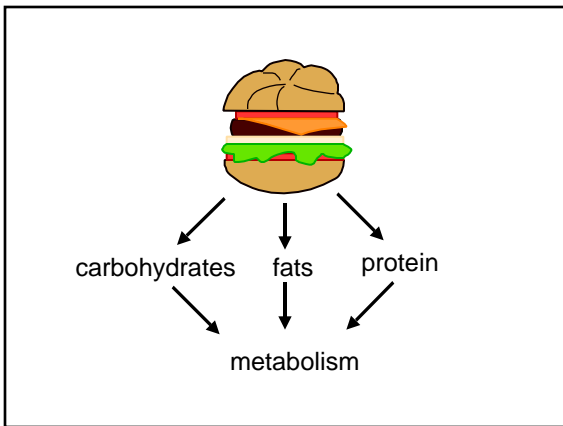
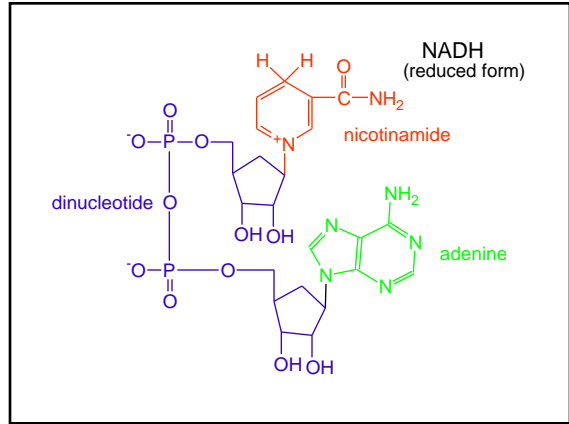
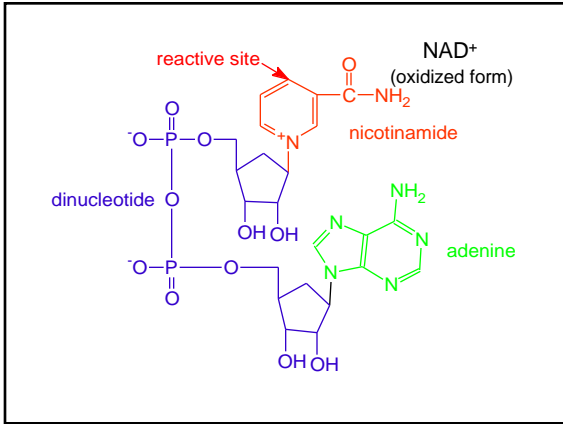
### Oxidation-Reduction Reactions

- Loss of electrons is **OXIDATION**.
  - Loss of energy
  - Loss of hydrogen
- Gain of electrons is **REDUCTION**.
  - Gain of energy
  - Gain of hydrogen

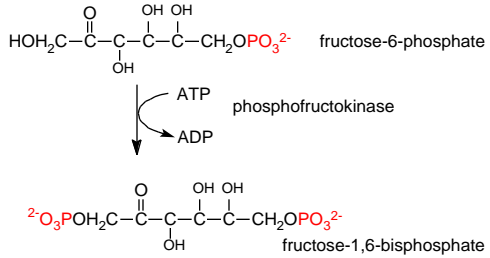
### Common electron acceptors

- Nicotinamide adenine dinucleotide (NAD<sup>+</sup>).
- Flavin adenine dinucleotide (FAD<sup>+</sup>)

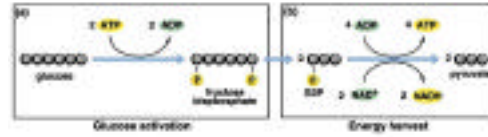




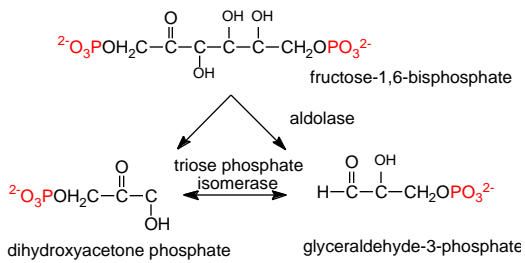
### Phosphofructokinase is the key enzyme in glycolysis



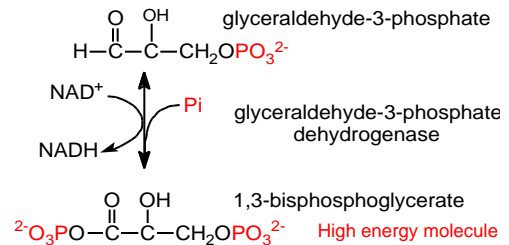
### Activating glucose



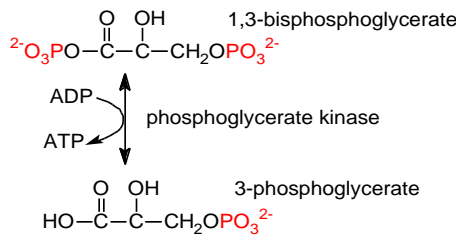
### Sugar-splitting



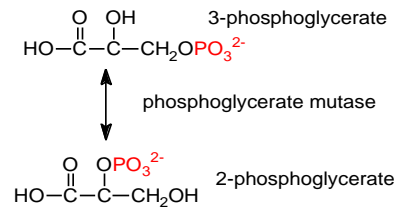
### Energy transfer



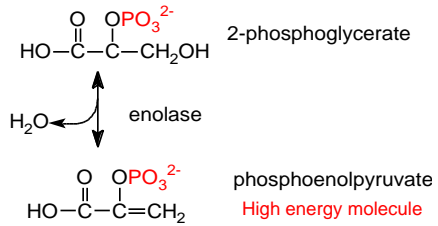
### Forming ATP (substrate phosphorylation)



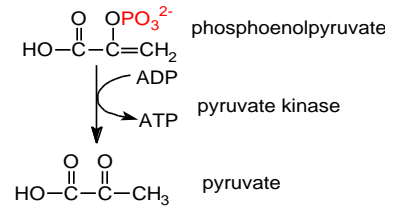
### Rearrangement



Dehydration increases the potential for transfer of phosphate to ADP



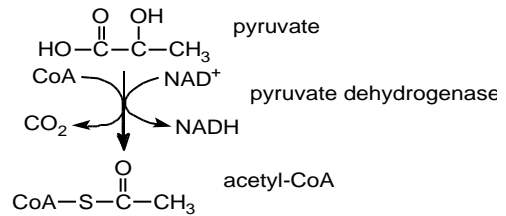
Transfer of P to ADP is irreversible (substrate phosphorylation)



### Glycolysis summary

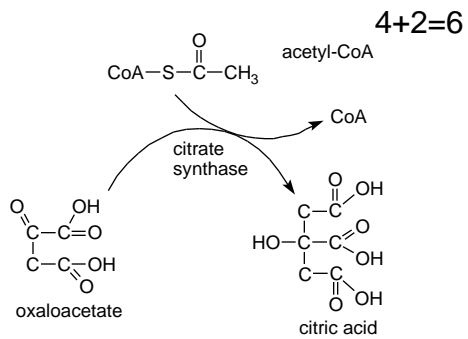
- |  |  |
|--|--|
| <ul style="list-style-type: none"> <li>■ IN</li> <li>■ 1 glucose</li> <li>■ 2 ATP</li> <li>■ 2 NAD<sup>+</sup></li> <li>■ 2 Pi</li> <li>■ 2 ADP</li> </ul> | <ul style="list-style-type: none"> <li>■ OUT</li> <li>■ 2 pyruvate</li> <li>■ 4 ATP</li> <li>■ 2 NADH</li> <li>■ 2 H<sub>2</sub>O</li> </ul> |
|--|--|

CoA activates the carbons from pyruvate

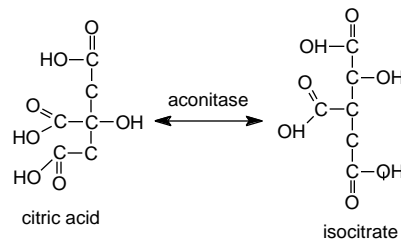


ATP is required to move pyruvate into the mitochondrial matrix

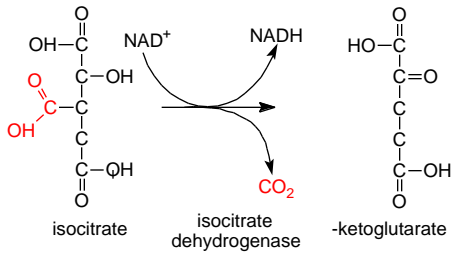
Acetyl-CoA can be used to synthesis fats



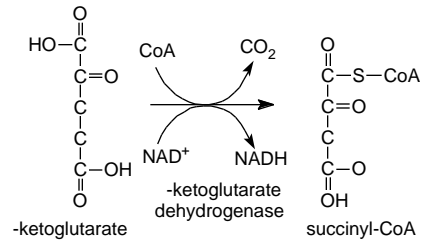
Moving an -OH



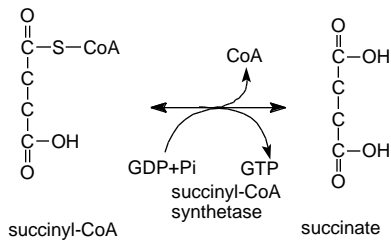
### Oxidation and decarboxylation



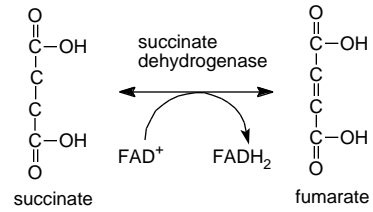
### Energizing the bonds



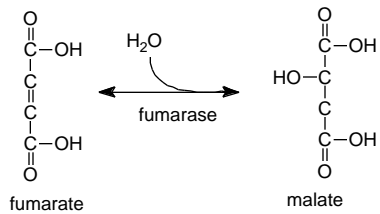
### Making ATP



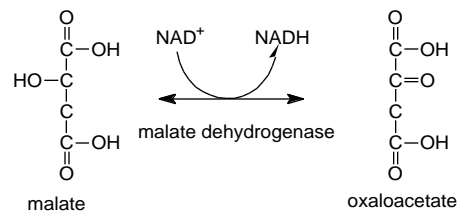
### Using FAD+



### Hydrating fumarate

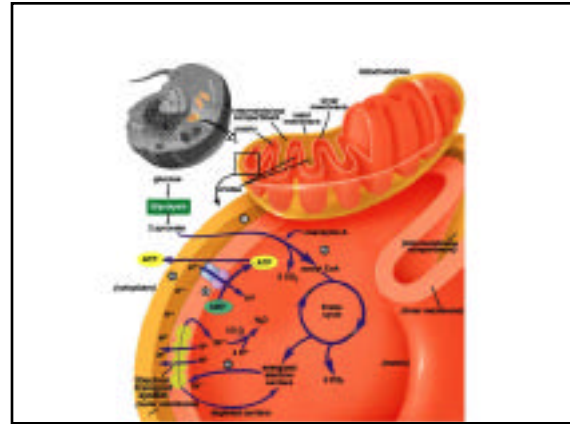


### Oxidation of malate

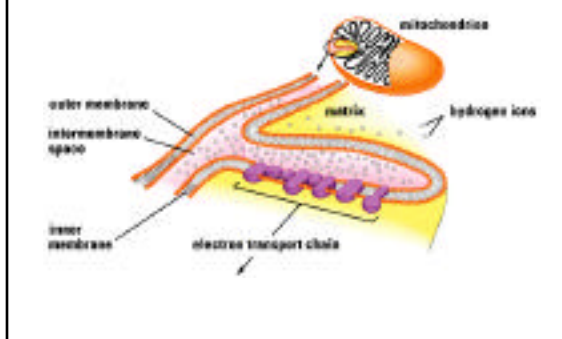


### Krebs cycle summary

IN	OUT
■ 2 acetyl-CoA	■ 2CO <sub>2</sub>
■ 6 NAD <sup>+</sup>	■ 6NADH
■ 2 FAD <sup>+</sup>	■ 2FADH <sub>2</sub>
■ 2GDP	■ 2GTP
■ 2Pi	■ 2H <sup>+</sup>
■ 2H <sub>2</sub> O	■ 2 CoA



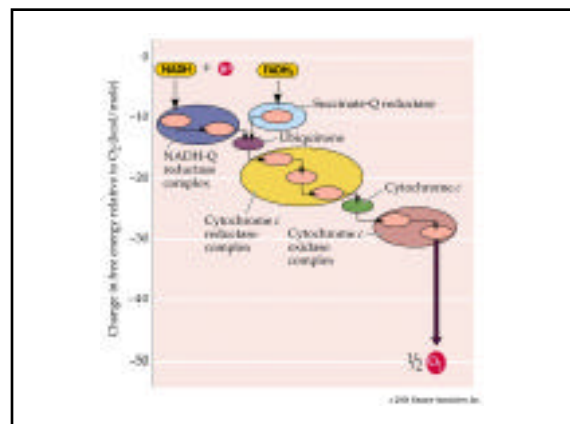
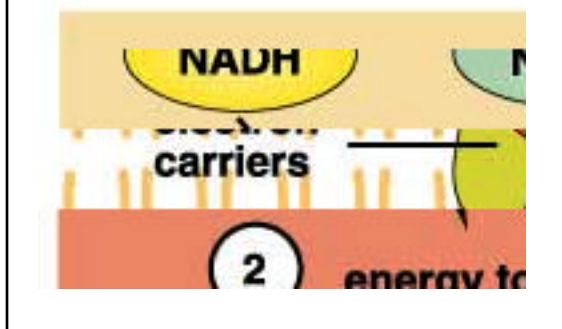
### Structure of the mitochondrion

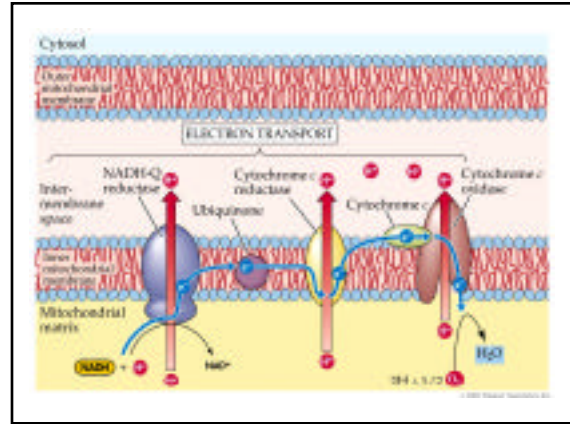
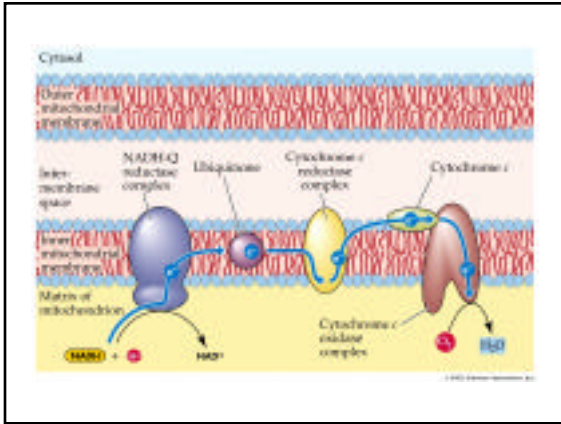


### Electron Transport System

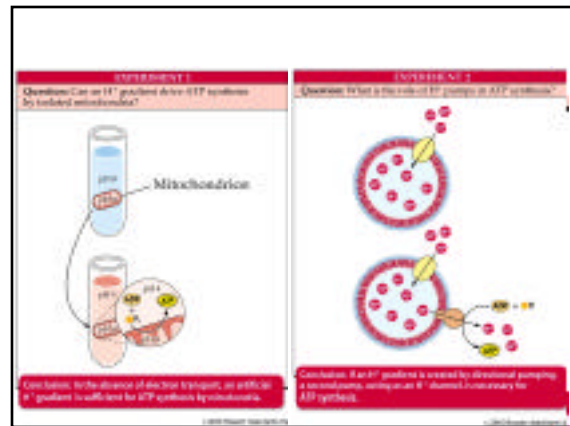
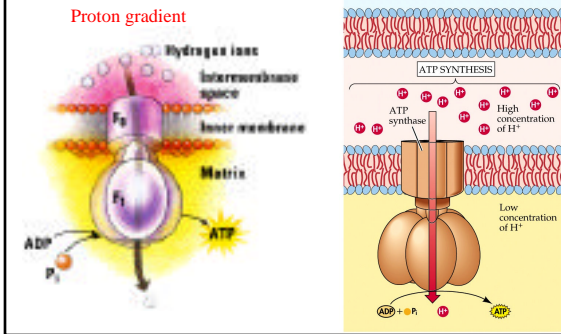


### Electron transport chain





ATP synthase uses energy from a proton motive force (chemiosmosis)



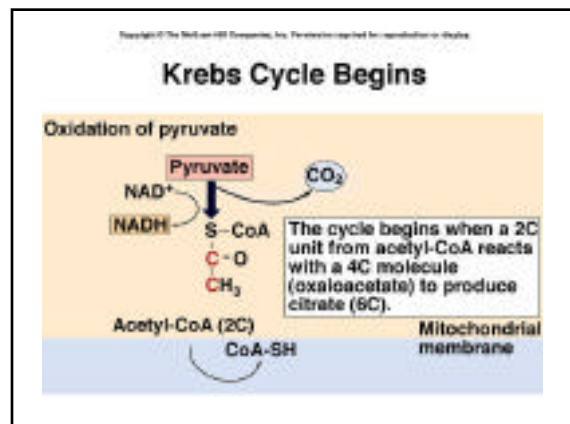
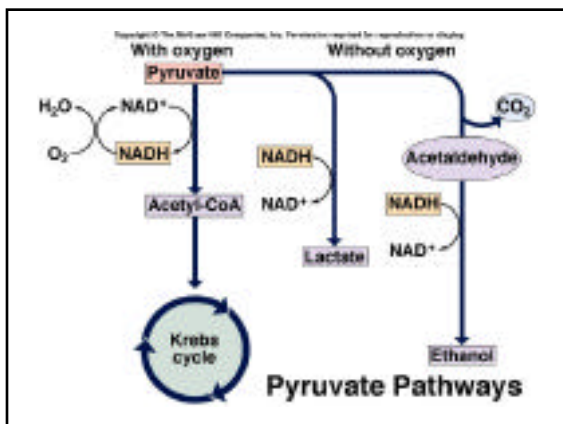
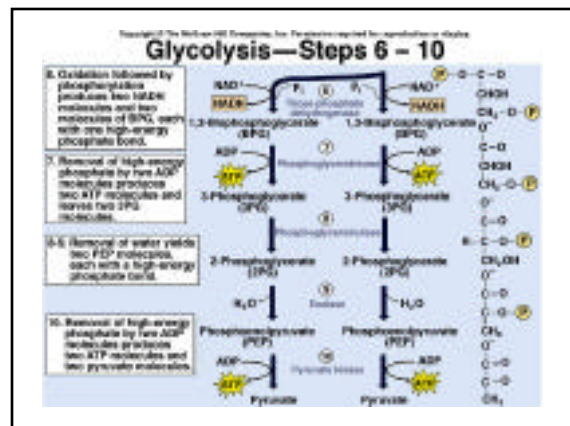
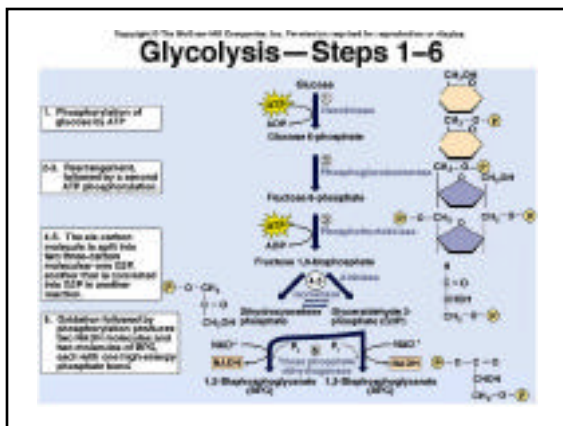
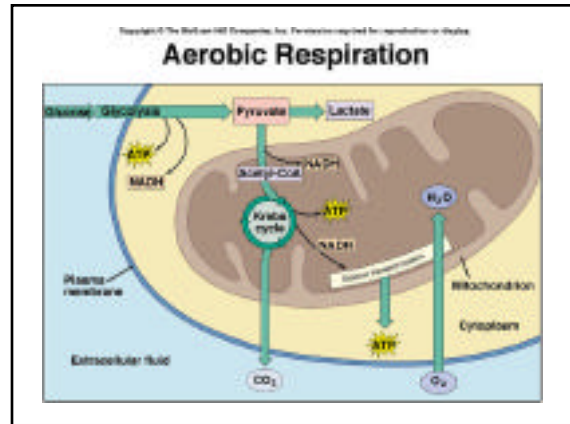
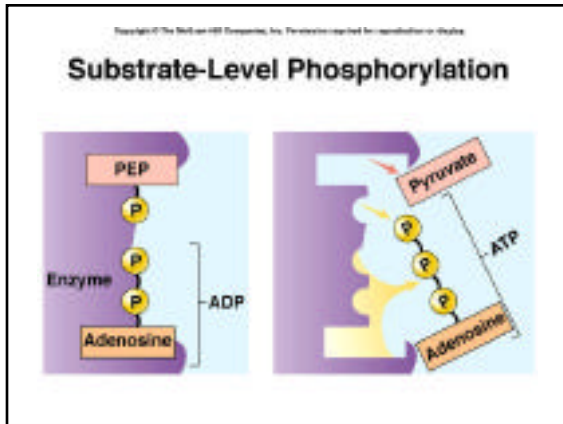
### Krebs cycle leads to 20 ATP

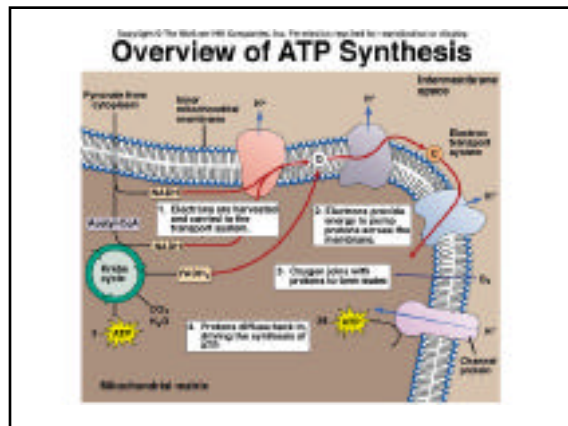
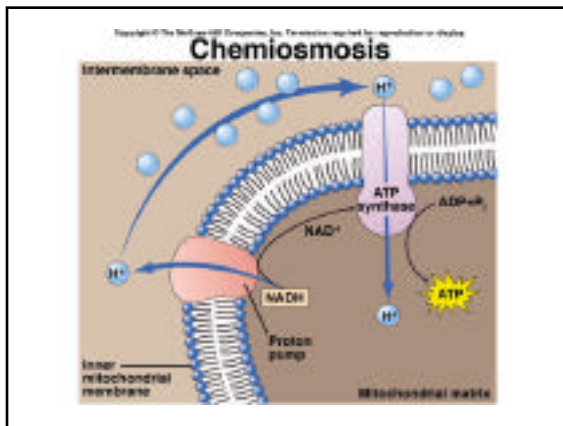
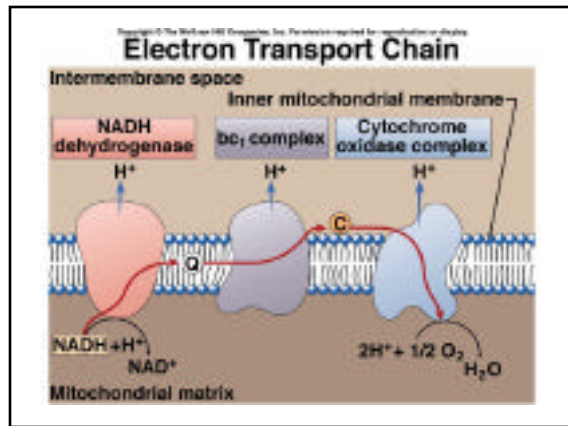
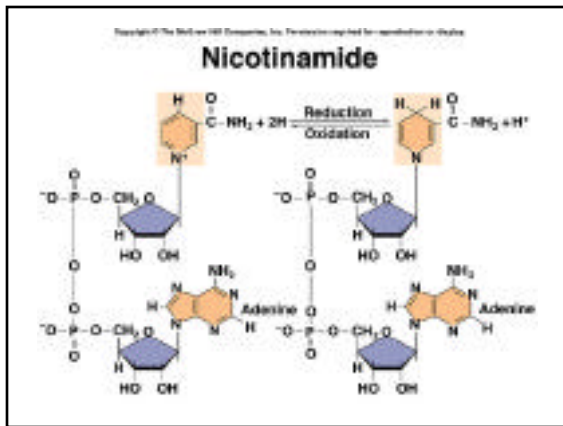
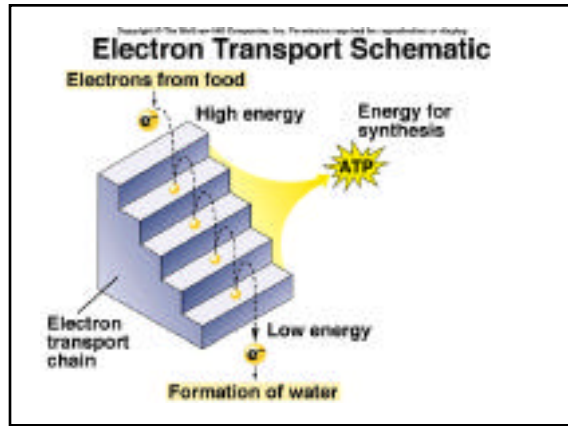
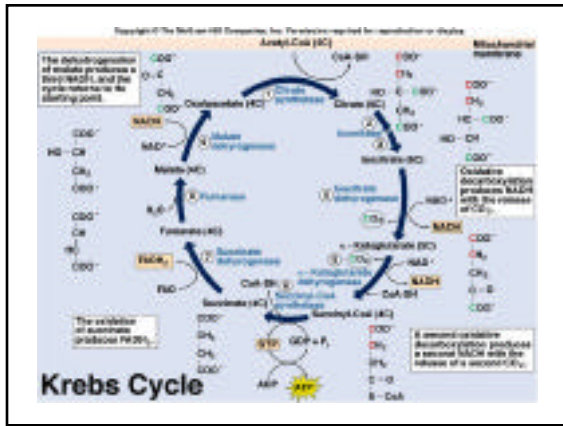
IN	OUT
■ 2 acetyl-CoA	■ 2CO <sub>2</sub>
■ 6 NAD <sup>+</sup>	■ 6NADH=15ATP
■ 2 FAD <sup>+</sup>	■ 2FADH <sub>2</sub> =3ATP
■ 2GDP	■ 2GTP=2ATP
■ 2Pi	■ 2H <sup>+</sup>
■ 2H <sub>2</sub> O	■ 2 CoA

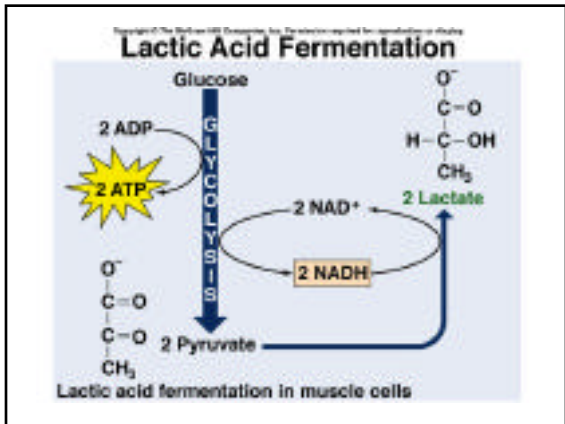
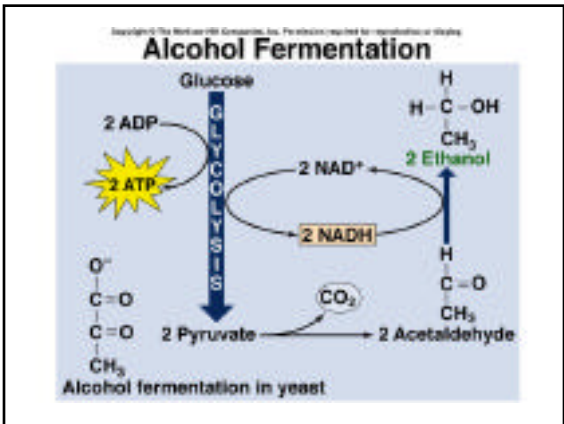
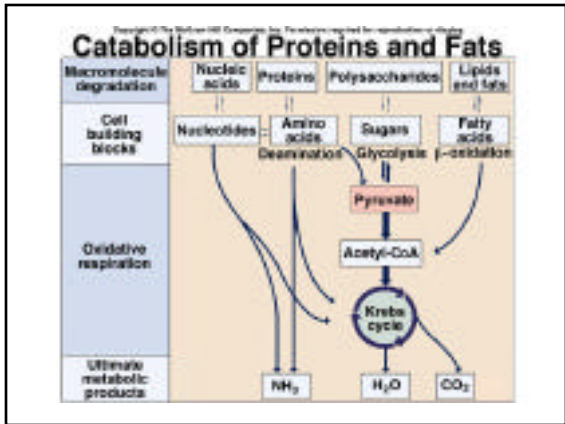
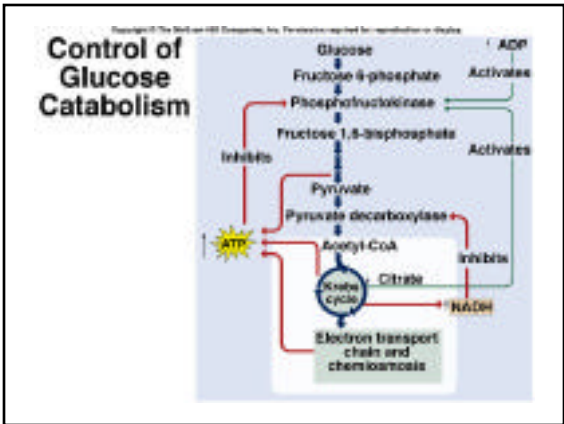
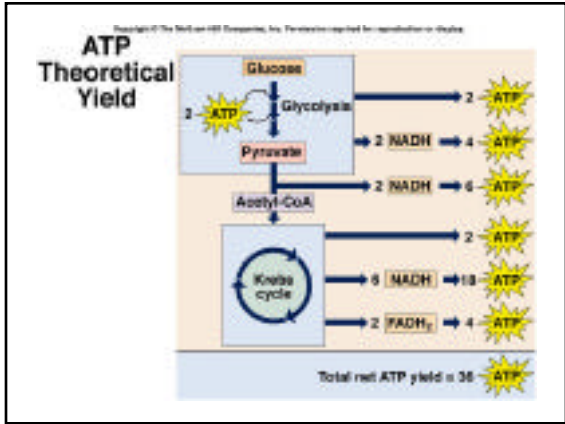
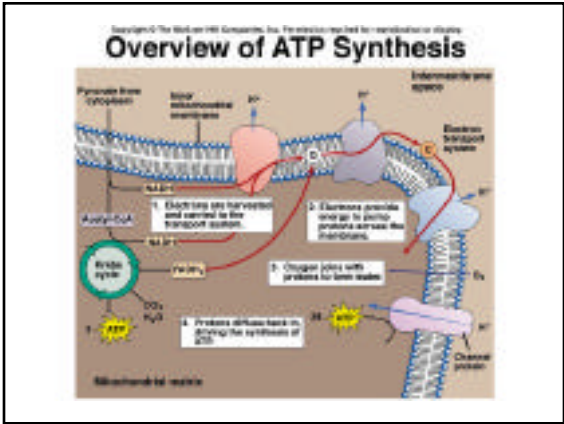
### Energy summary

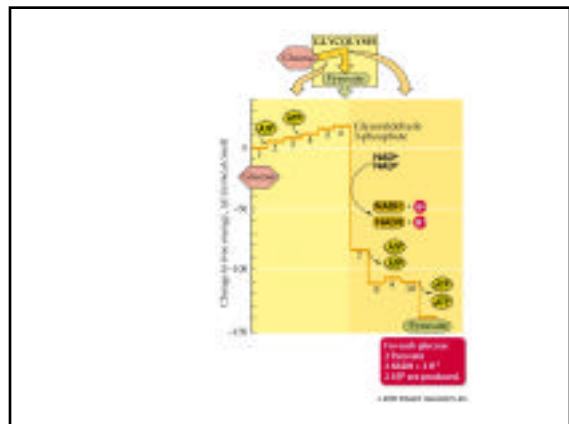
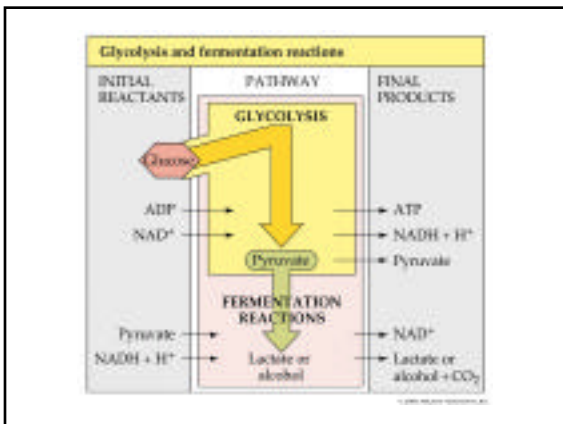
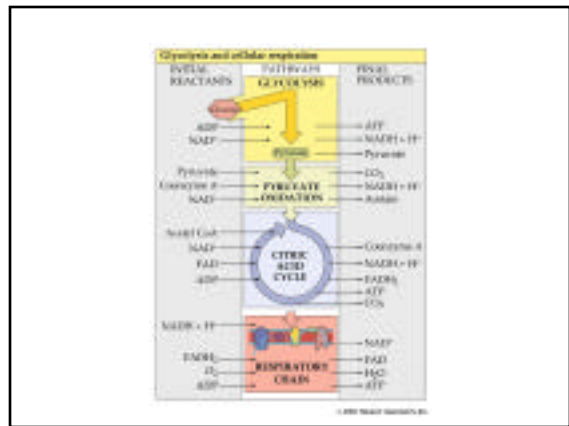
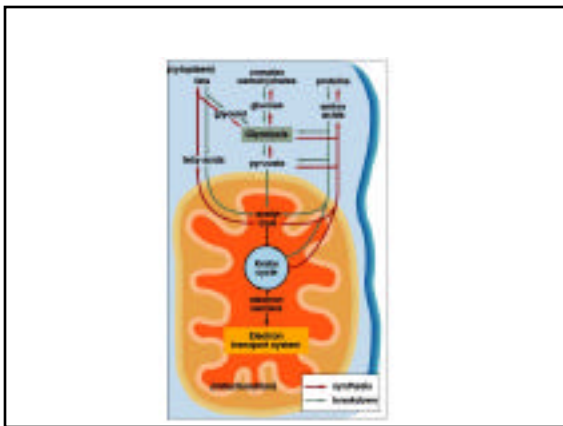
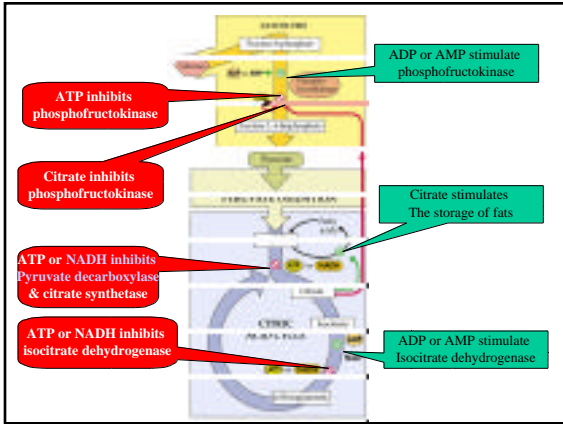
■ Glycolysis	■ Glycolysis
■ 2ATP	■ 7 ATP
■ 2NADH	■ Pre-Kreb's cycle
■ Pre-Kreb's cycle	■ 5 ATP
■ 2NADH	■ Kreb's cycle
■ Kreb's cycle	■ 20 ATP
■ 6NADH	
■ 2FADH <sub>2</sub>	
■ 2GTP	
	NADH = 2.5 ATP
	FADH <sub>2</sub> = 1.5 ATP
	Total ATP = 32

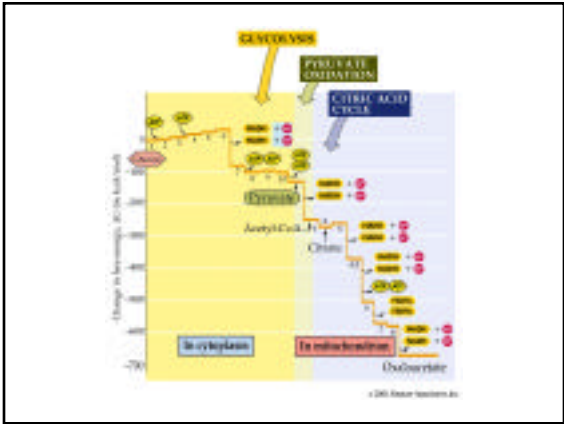












### 7.1 Cellular Locations for Energy Pathways in Eukaryotes and Prokaryotes

EUKARYOTES	PROKARYOTES
<b>External to mitochondrion</b>	<b>In cytoplasm</b>
Glycolysis	Glycolysis
Fermentation	Fermentation
	Citric acid cycle
<b>Inside mitochondrion</b>	<b>On inner face of plasma membrane</b>
Inner membrane	Pyruvate oxidation
Pyruvate oxidation	Respiratory chain
Respiratory chain	
Matrix	
Citric acid cycle	

© 2011 Sinauer Associates, Inc.

